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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/637,149	08/08/2003	Gerald E. McDonnell	MEDZ 2 01304	3426
7590 05/15/2007 Thomas E. Kocovsky, Jr. FAY, SHARPE, FAGAN, MINNICH & McKEE, LLP			EXAMINER	
			HORNING, MICHELLE S	
Seventh Floor 1100 Superior Avenue		ART UNIT	PAPER NUMBER	
-	Cleveland, OH 44114-2518		1648	
	-			
		•	MAIL DATE	DELIVERY MODE
		·	05/15/2007	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)			
	10/637,149	MCDONNELL ET AL.			
Office Action Summary	Examiner	Art Unit			
	Michelle Horning	1648			
The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply					
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DATE of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. If NO period for reply is specified above, the maximum statutory period we failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be time vill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).			
Status					
Responsive to communication(s) filed on <u>01 Mar</u> This action is FINAL . 2b) ☑ This Since this application is in condition for alloward closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro				
Disposition of Claims					
4) Claim(s) 1-8 and 22-29 is/are pending in the ap 4a) Of the above claim(s) 2-4 and 14 is/are with 5) Claim(s) is/are allowed. 6) Claim(s) 1, 5-13, 15-18 and 22-2 is/are rejected 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restriction and/or	drawn from consideration.				
Application Papers					
9) The specification is objected to by the Examiner 10) The drawing(s) filed on is/are: a) access applicant may not request that any objection to the Replacement drawing sheet(s) including the correction of the output of	epted or b) objected to by the Edrawing(s) be held in abeyance. See ion is required if the drawing(s) is obj	e 37 CFR 1.85(a). ected to. See 37 CFR 1.121(d).			
Priority under 35 U.S.C. § 119					
 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) All b) Some * c) None of: 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received. 					
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal Pa 6) Other:	nte			

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DETAILED ACTION

This office action is responsive to communication filed 3/1/2007. The status of the claims is as follows: claims 1, 5-13, 15-18 and 22-29 are under current examination, claims 2-4 and 14 are withdrawn from consideration as drawn to non-elected inventions and claims 19-21 are canceled.

The Office acknowledges that Applicant has corrected the specification by deleting all references related to Figure 6. Previously, a drawing for figure 6 was not included in the instant specification.

Claim Rejections-Withdrawn

The following claim rejections have been withdrawn due to claim amendments.

- 1. 35 USC 112, 2nd paragraph;
- 2. 35 USC 102; and
- 3. 35 USC 103 (Ernst and Race, 1993 and Werner et al, 1983).

The following claim rejection has also been withdrawn.

4. 35 USC 103 (Ernst and Race, 1993, Werner et al, 1983, Cooper, 1913 and Yamamoto et al, 2001). This rejection has been withdrawn not due to any persuasive arguments presented by the Applicant but in order to modify this rejection to include additional references to provide a more thorough explanation for the Applicant. Of note, the Examiner is well aware that a protein is not a bacterium.

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Claim Rejections-New

35 U.S.C. 103(a)

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

Claims 1, 4-13, 15-18 and 22-29 are rejected under 35 U.S.C. 103(a) as being unpatentable over Prusiner (1982), Gassett et al (1993), Nandi et al (2002), Cai et al (2002), Ernst and Race (1993, previously cited) and US Patent Application 10/467591 (hereinafter as "Kritzler et al", 2002). The limitations of the claims are as follows:

1. a method of treating a body which is contaminated with prions by contacting the body with a composition comprising a phenol and a soluble inorganic salt to inactivate the prions;

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- 2. wherein the phenol is up to 0.2M;
- 3. wherein the phenol has a log Pc value of 2 to 6.5, 2-5 and at least 4;
- 4. wherein the concentration of phenol is at least 10%;
- 5. wherein the soluble inorganic salt includes NaCl;
- 6. wherein the phenol includes o-phenylphenol and o-benzyl-p-chlorophenol;
- 7. wherein the phenol complexes with the prions and precipitates;
- 8. wherein the phenol has minimal solubility;
- 9. wherein the body includes a surface;
- 10. wherein the soluble inorganic salt is at a concentration of up to 5%;
- 11. wherein the composition further comprises a surfactant, more specifically, dodecylbenzene sulphonic acid;
 - 12. wherein the composition further comprises an acidic sequestering agent; and
- 13. a method of treating a body contaminated with prions comprising contacting the body with a composition to inactivate prions, the composition comprising a phenol, co-solvent, water and a surfactant from the group of suphonic acids, sulfonates and combinations thereof.

Prusiner reviews six distinct treatments of a scrapie agent that leads to its inactivation, including treatment by phenol and chaotropic salts (see page 138). This reference discloses that "extraction with phenol, a potent denaturant of protein, under various salt and pH conditions destroyed infectivity" of the scrapie agent (see page 139). Prusiner concludes "denaturation of a protein within the scrapie agent leads to inactivation of the infectious particle". Thus, both phenol and salt have been shown to

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denature (alter the structure of) the scrapie agent that affects its function (infectivity). It is further noted here that phenol and salt have been combined with denaturing effects, as quoted above. This reference does not disclose the detailed use of NaCl or of ophenylphenol or o-benzyl-p-chlorophenol in order to inactivate prion proteins. This is found in the prior art as discussed below.

Gassett et al discloses multiple conditions, including increases in ionic strength, which may perturb infectivity of PrP 27-30 by disruption of its secondary structure. This reference points out that varying concentrations of NaCl alters the conformation stability of PrP 27-30 caused by the perturbation of multiple molecular forces in the maintenance of polymers (page 3-4). For further elucidation of this discussion, the Nandi et al reference is addressed here. This reference discusses the unfolding of the prion protein in various salt solutions at neutral pH (see entire document). Salt solutions, as disclosed by this reference, have large effects on the structure and properties of proteins, including their solubility, denaturation and activity (see page 11020-1). This reference also provides the following recitation: "A considerable amount of studies on the effects of salts on the structural properties of proteins have been carried out in the past which suggest that at least two effects of salt, viz., their effects on solvent (water) structure and electrostatic interaction with charged groups of the protein, make major contributions to the structure-stabilizing properties of the proteins" (see page 11021)... Thus, it is well known in the prior art that salt, more specifically NaCl as taught by Gassett et al and Nandi et al, can denature prion proteins. In addition to the Gassett et al and Nandi et al, Cai et al teach the solvent-dependent precipitation of prion protein in

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various pH, salt and ethanol concentration (see Abstract). High salt solutions (0.25M) facilitated precipitation of both scrapie and cellular prion proteins (see page 31). According to Cai et al, the salt content plays a "critical role in determining protein interactions in solution (page 34). Given the teachings above, it would have been obvious to one of ordinary skill in the art to alter the NaCl concentration of a composition in order to achieve optimal disruptive results of secondary structure. The above references do not teach the specific use of o-phenylphenol and o-benzyl-p-chlorophenol in inactivating prions.

Ernst and Race teach a method in which the scrapie agent of brain homogenates is inactivated following treatment with LpH, or an aqueous phenolic disinfectant comprising both o-phenylphenol and o-benzyl-p-chlorophenol (see page 196). While this reference teaches using a concentration of 90% of LpH (page 197) which equates to 9% of phenolic derivative concentration (see 198 for conversion), varying the phenolic concentration would have been obvious to one of ordinary skill in the art in order to achieve optimal results of prion inactivation. Of note, the partition coefficient is dependent on the concentration value. The references above, however, do not teach a method using a composition that further comprises dodecylbenzene sulphonic acid.

Kritzler et al teach a method and a composition for treating a surface contaminated with a scrapie prion protein. The composition comprises one or more agents which favor the conformational unfolding of a scrapie prion protein (see Abstract), including inorganic salts and surfactants (see paragraphs 41 and 42).

Dodecyl benzene sulfonate is disclosed in paragraph 41 as a denaturant that tends to

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"bind to proteins and initiate unfolding of tertirary structure". According to the instant specification, dodecylbenzene suphonic acid can be used as either a surfactant or as an acidic sequester agent. Further, Kritzler et al describe the use of a cosolvent, including m-Cresol (paragraph 39). A cosolvent is defined by the instant specification in paragraph 45 and includes a polyol which comprises only carbon, hydrogen and oxygen atoms. M-Cresol fit this definition. Kritzler et al disclose that such solvents "tend to denature, dissolve or swell proteins. Generally the products are not completely unfolded and possess an ordered conformation which differs from the native state" (paragraph 39).

Thus, it would have been obvious to one of ordinary skill in the art to combine the teachings of the references above in order to achieve a composition or a method of using such a composition comprising multiple agents that would alter the structure of the prion protein. One would have been motivated to combine various agents in order to denature the prion protein because as suggested by Kritzler et al "Many proteins are prone to loose their natural three dimensional folding pattern ("secondary and tertiary structure") and to become "denatured". The denaturation includes breakdown of the intramolecular interaction, especially hydrogen and disulphide bonds, and thus the loss of the secondary structure which virtually all native proteins have in at least parts of the molecule, and which generally is decisively responsible for the activity of the protein" (paragraph 25). Given the combined teachings of the

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references above show that phenols, salts, dodecyl benzene sulfonate and polyol, such

as m-Cresol, alter the structure of a prion protein, there would have been a reasonable

expectation of success in the inactivation of the prion protein (see Prusiner discussion

above). The invention as a whole was clearly prima facie obvious to the ordinary artisan

at the time the invention was made.

CONCLUSIONS

No claim is allowed.

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to Michelle Horning whose telephone number is 571-272-

9036. The examiner can normally be reached on Monday-Friday, 8:30 am to 5 pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Bruce Campell can be reached on 571-272-0974. The fax phone number for

the organization where this application or proceeding is assigned is 570-272-8300.

Information regarding the status of an application may be obtained from the

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866-217-9197 (toll-free).

Michelle Horning

Patent Examiner

BRUCE R. CAMPELL, PH.D SUPERVISORY PATENT EXAMINER

TECHNOLOGY CENTER 1600